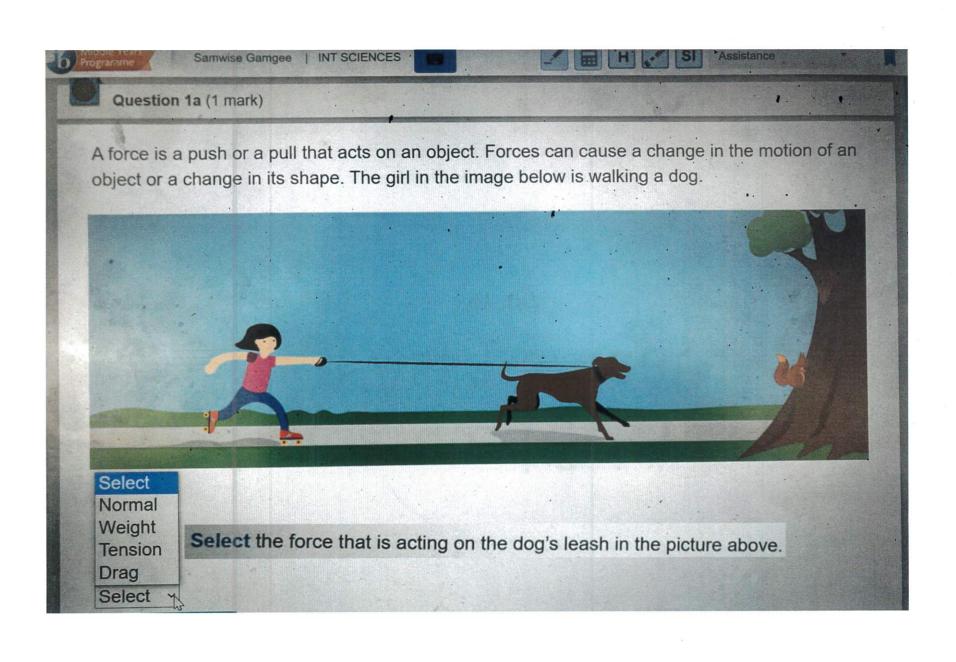
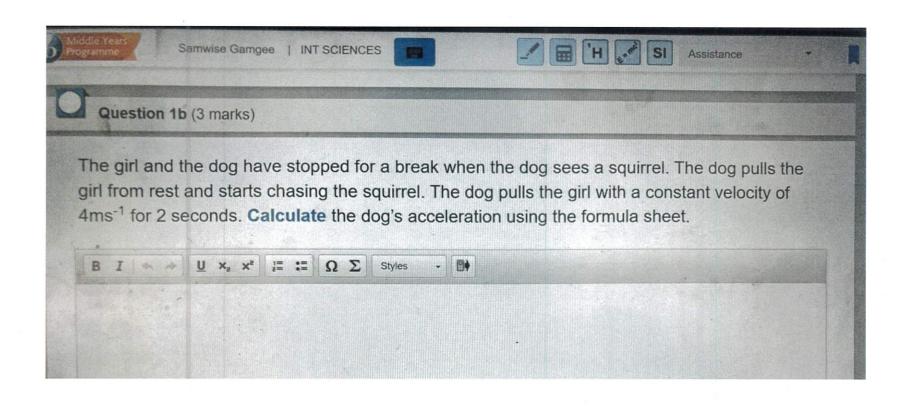
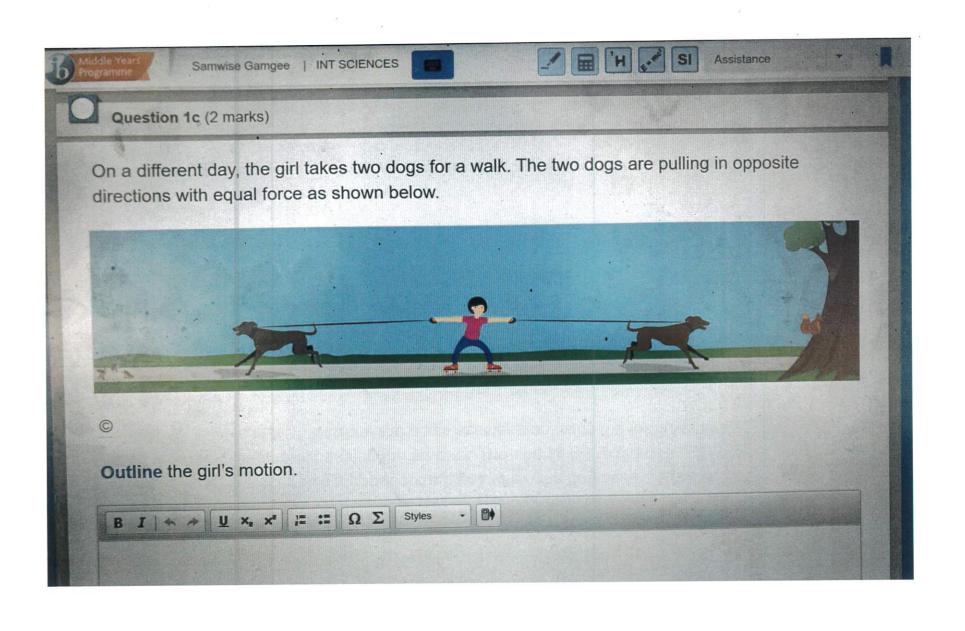
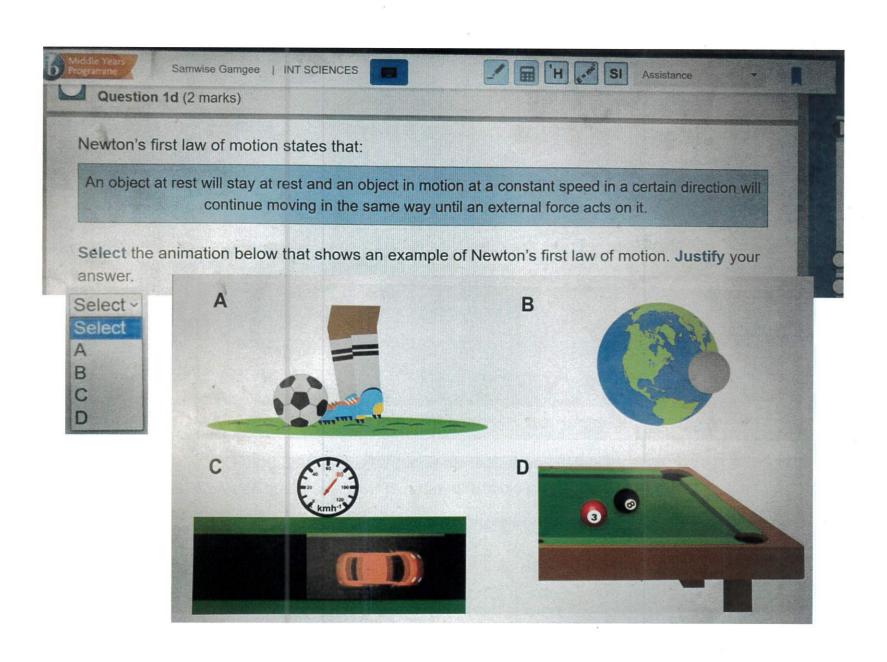
Integrated Science e-Assessment

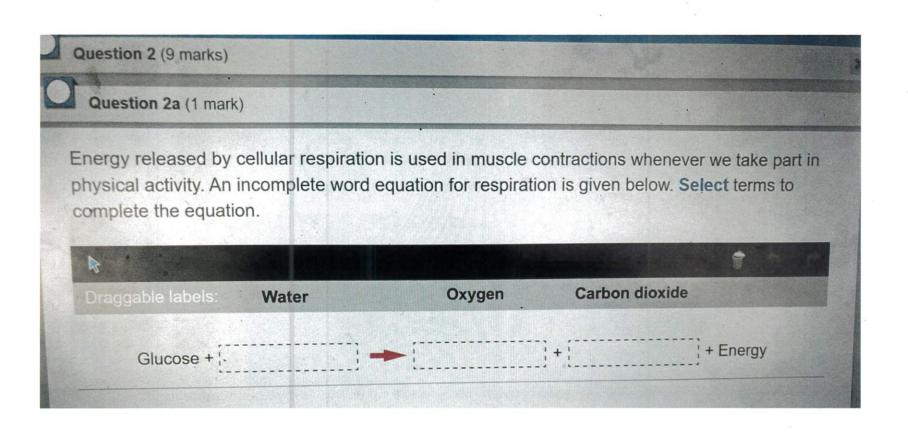
Nov 2020

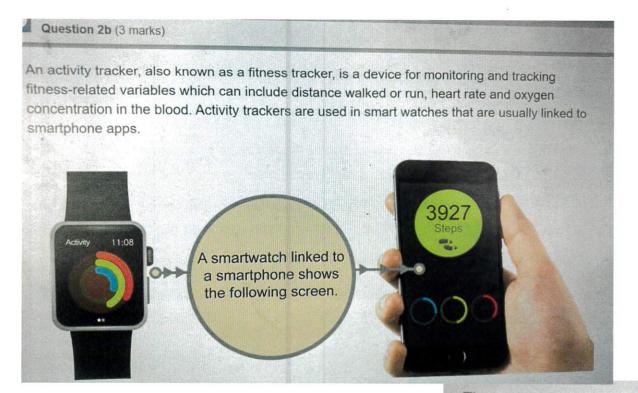


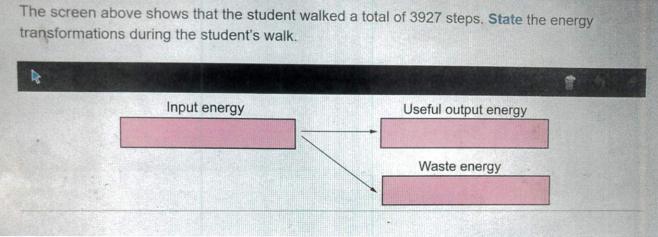


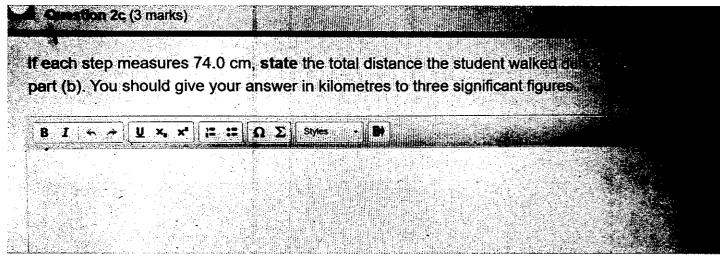


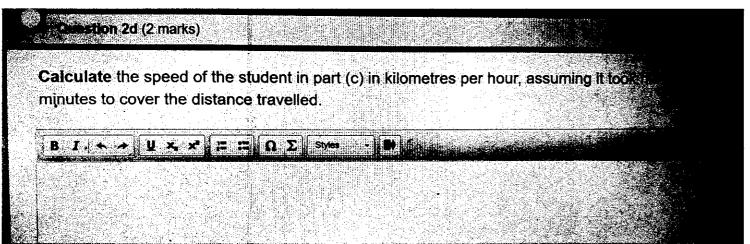


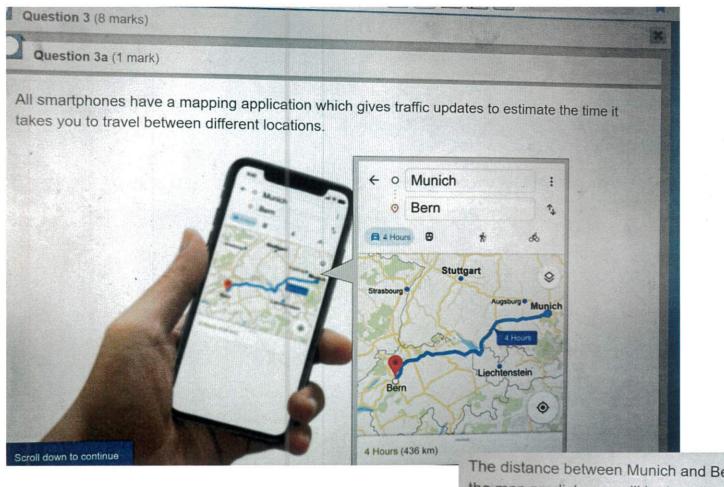






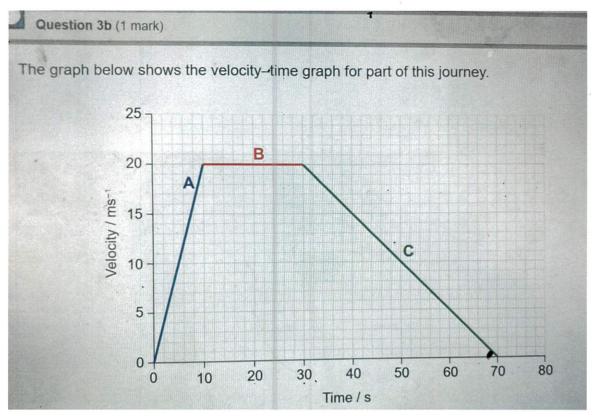


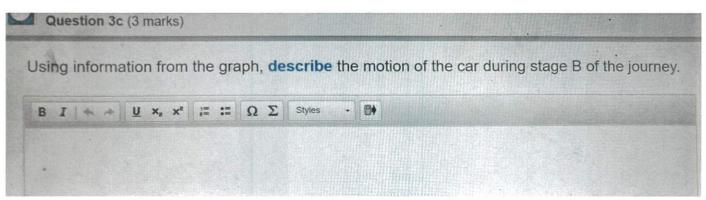


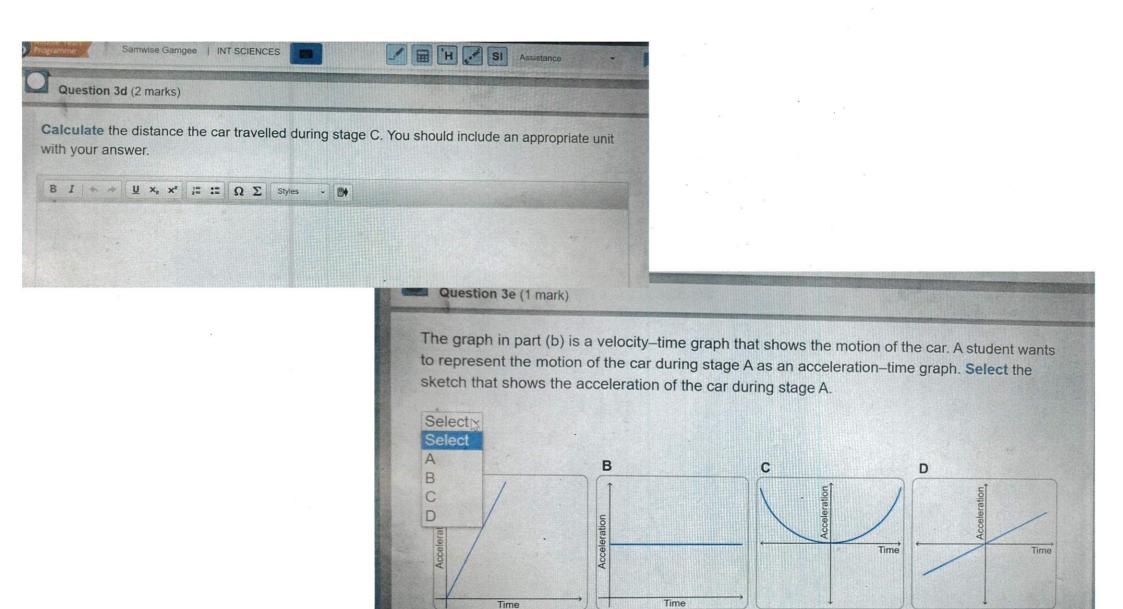


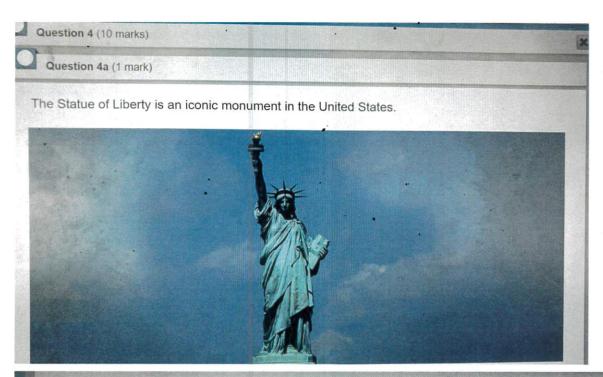
The distance between Munich and Bern is 436 km. Using the map above, calculate the speed the map predicts you will be travelling by car. Show your answer in kilometres per hour.











The Statue of Liberty is made from hundreds of copper sheets on a framework of steel. When the statue was first built it was the red-orange colour of copper, but as the copper slowly reacted with oxygen in the atmosphere, the red-orange colour changed to the familiar green colour we see today.

Select the area of the periodic table where copper is found.

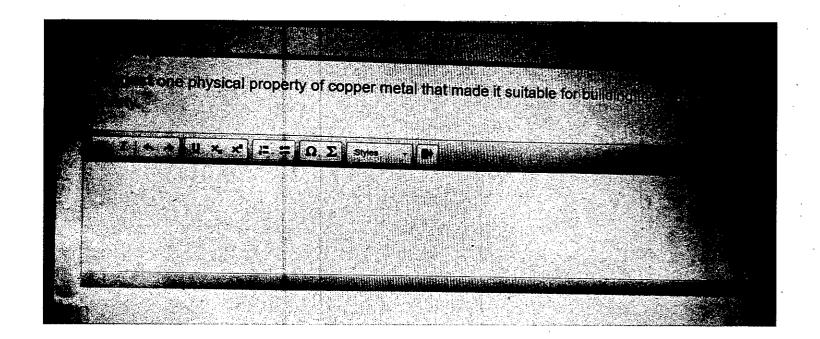
Select

Select

Group 1 – alkali metals

Group 2 – alkaline earth metals

Transition metals
Group 7 – halogens



Question 4c (2 marks)

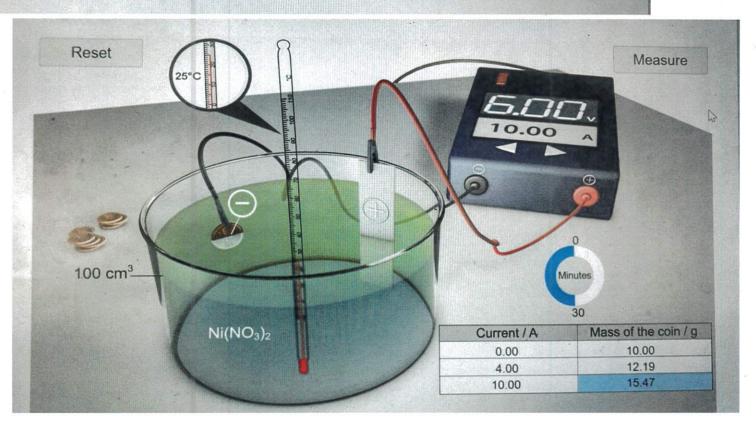
The Statue of Liberty changed colour because the copper metal reacted with oxygen in the atmosphere in a process called corrosion. One of the methods used to prevent corrosion is electroplating. Electroplating is a process that involves coating a thin layer of metal on the surface of another metal using an electric current. This process is commonly used in industry to prevent corrosion or to cover a relatively cheap metal with a more expensive metal, such as silver, gold or zinc. You can see many electroplated objects around you, for example: silver-plated tableware, watches and rings.

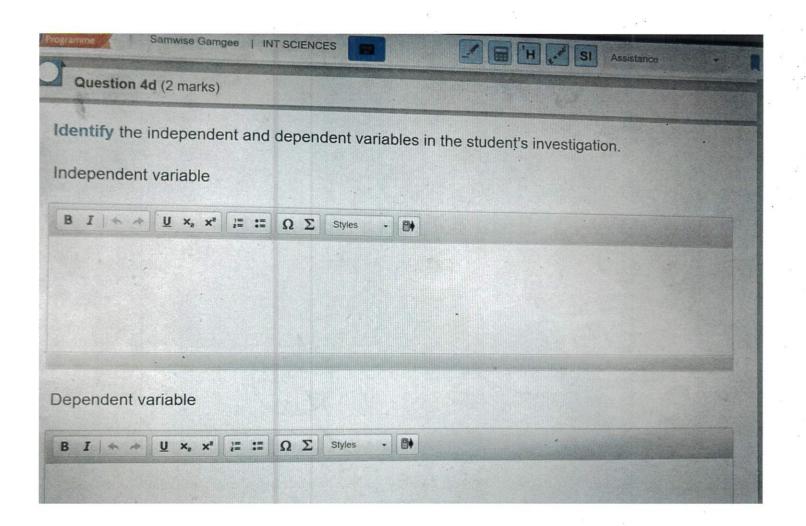
The animation below shows how a copper coin can be electroplated.

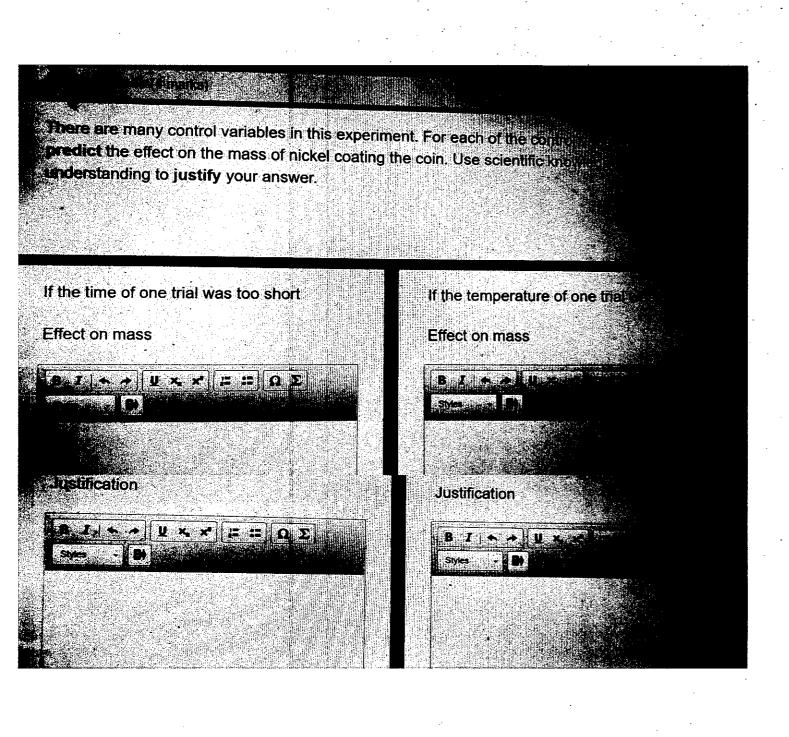


There are several factors that could affect the mass of the electroplated metal on the coin, for example: current, voltage, concentration of solution, temperature, surface area of metal coin in solution and duration of electroplating.

A student wanted to investigate the electroplating process in her school laboratory. She wanted to coat a copper coin with nickel using a 10 g nickel bar as a source. She wanted to test how changing the current will affect the mass of nickel coating the copper coin. Her equipment is shown below:





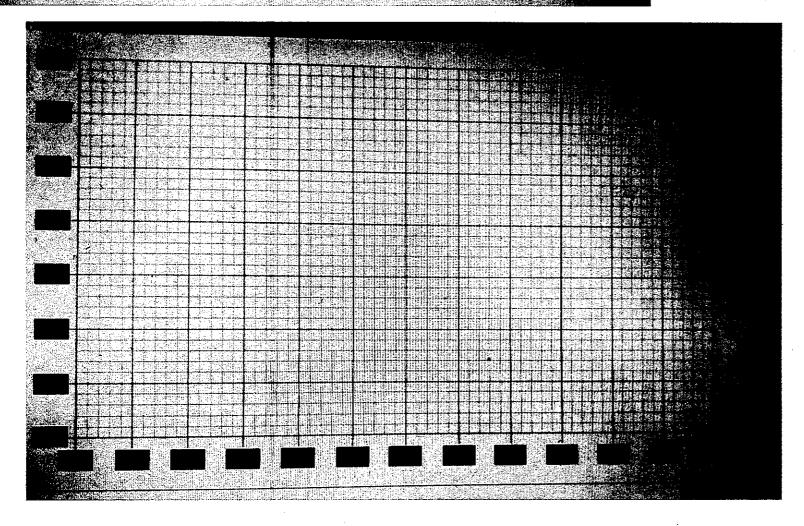


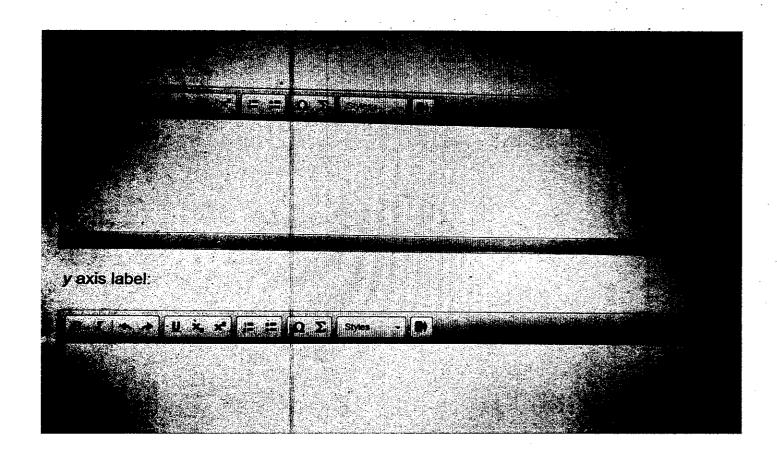
The Student continued her investigation by processing and interpreting her the mass of nickel coating the coin.

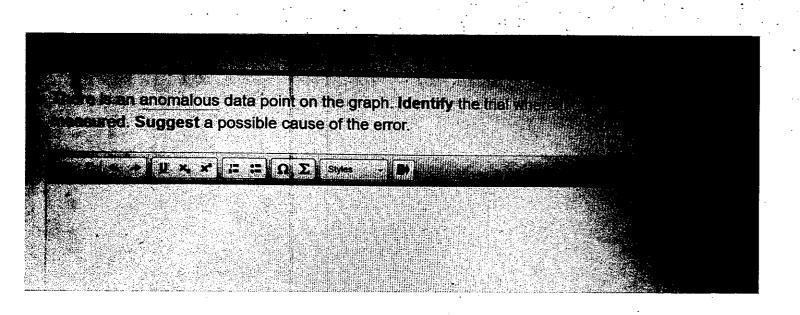
• 5 •	A fe. chompatie	able below:		
Avine	Mass of nickel coating the coin / g			in the second
	Trial 1	Tnal 2	Trial 3	
0.00	0.0	0.0	0.0	
2.00	1.1		1.0	
4.00	2.2	21	2.2	
6.00	33	33	3.1	733.25 P
8.00	4.4	7.6	4.5	
	5.5	54	5.4	
	6.6	6.4	6.3	

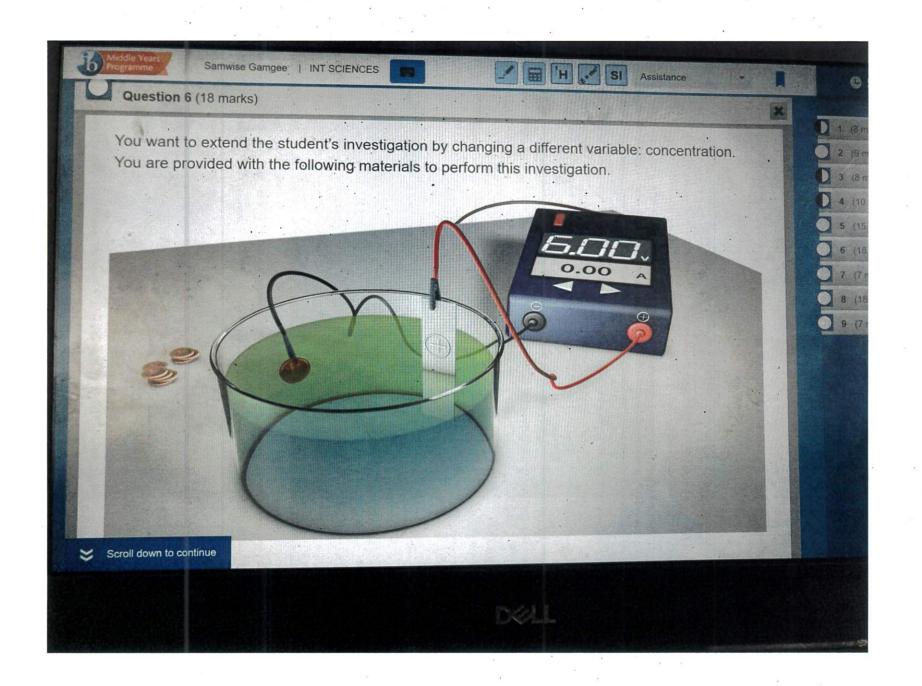
Recomplete. Calculate the missing average and low your working in the response box below.

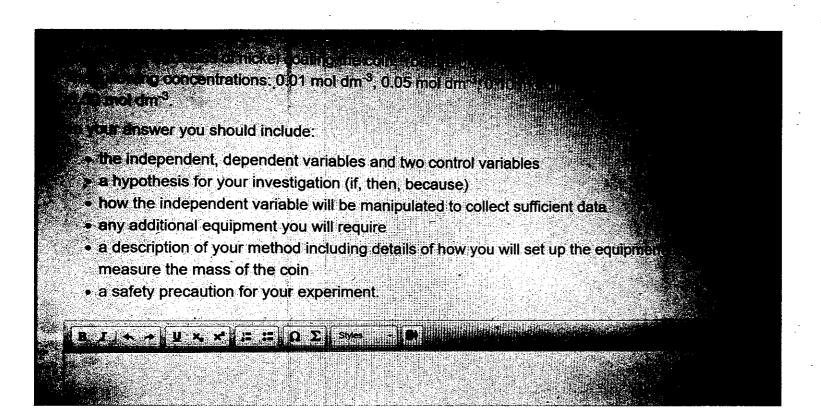
c-sent the average data from part (a) in a graph. Add labels for the axes



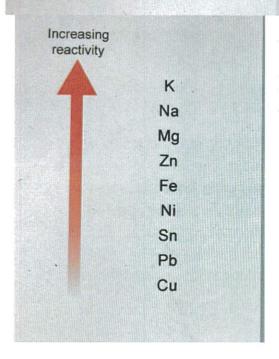


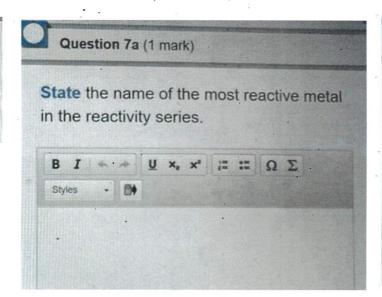


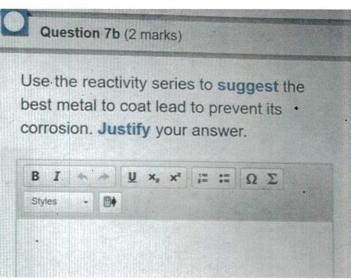


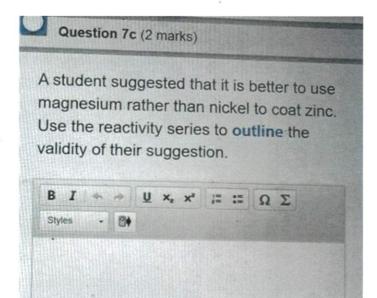


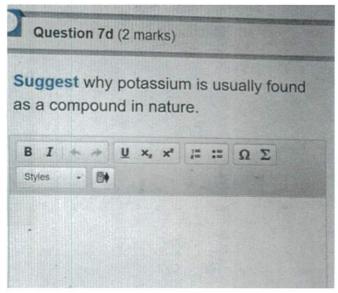
The reactivity series lists metals in order of increasing reactivity starting with the least reactive metals at the bottom. The most reactive metals will corrode most quickly.

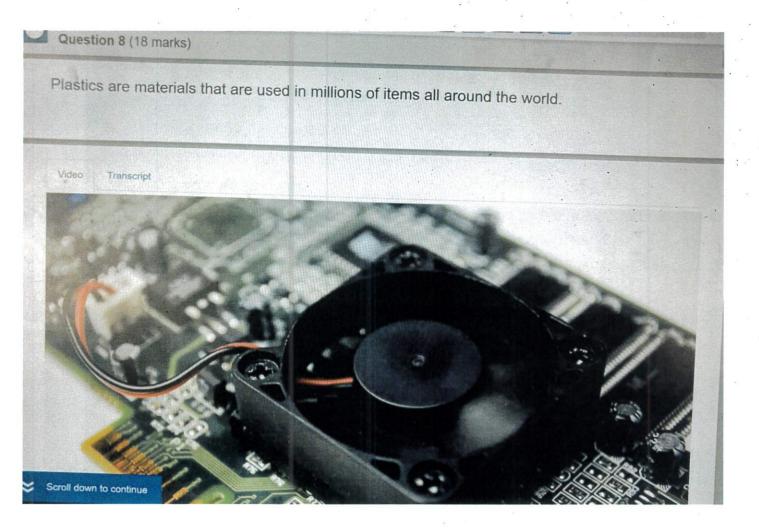












See next page for video transcript

are versatile materials that are used in millions of items all around the action tonnes of plastics were made with a total export value of 72.7 billion US cales sed because they are very strong and long-lasting, but these properties are the cales of the cales are the cales of t

Priven plastic items reach the end of their useful life, 55 % are discarded, 20 % are items remaining 25 % of plastic waste is burnt. Much of this discarded plastic ends up in our

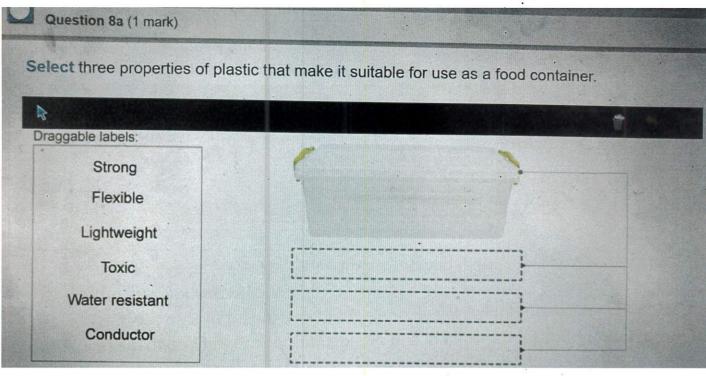
we cean, the plastic waste can affect marine life in different ways: It can be come tangled around marine animals.

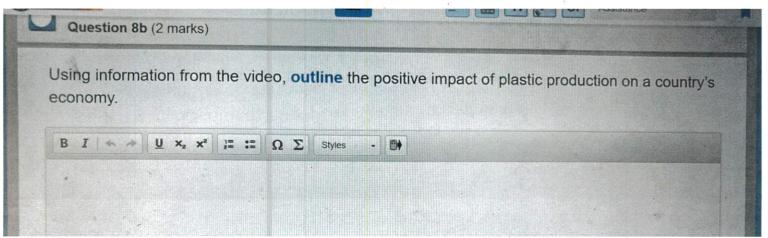
testitionally, discarded plastics have been left to break down in landfill sites or they are to be a suffice are burned they release dangerous chemicals such as hydrochloric acid, suffice the pollutants.

Surrently, scientists are researching solutions to deal with plastic waste. One of the propession of t

ிற் compounds such as carbon dioxide, water and methane can be கவ லிருந்திரு carbon cycle.

्रह्मित्रा use of microorganisms would take place under san ा विकास के a waste plastic collection site and analysis un ा किन्द्रीमांक्व for them to grow







Rice plastic is <u>discarded,</u> it takes many hundreds of years to break down. Micros Might offer a faster process to increase the rate of breakdown.

Discuss and evaluate the environmental and economic implications of the industric microorganisms to break down plastics. In your answer you should include:

- why plastic pollution is a problem that needs to be solved
- advantages to the environment of the use of microorganisms to break down plastics
- disadvantages to the environment of the use of microorganisms to break down plastics
- economic considerations
- a concluding appraisal comparing this method to the traditional method of burning plass

